

Syngas generation via pyrolysis-dry reforming of plastics over Ni-Mo₂C/ γ -Al₂O₃ catalyst - the effect of amine on structural and catalytic properties

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Keywords: catalysts, hydrogen, plastic waste, CO₂ utilization.

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In response to pressing global challenges, such as climate change, unmanaged plastic waste, and dwindling fossil resources, prioritizing sustainable development is imperative (Evode et al., 2021; Höök and Tang, 2013; Li et al., 2021). Introducing innovative, eco-friendly technologies to industries is key to securing a brighter future for the next generations. Particular attention should be paid to sustainable technologies that would allow the recovery of resources or energy from waste. Among these, pyrolysis combined with dry reforming (PCDR) of plastic waste stands out, garnering increasing interest (Olazar et al., 2024; Pawelczyk et al., 2022; Saad and Williams, 2017). PCDR involves decomposing plastic feedstock thermally under inert gas, followed by reforming the resulting volatiles to produce value-added products, e.g. synthesis gas. This approach not only tackles plastic waste management, but also offers an alternative to non-renewable resources for hydrogen production. Compared to conventional plastic waste management methods - incineration or landfilling, which emit toxic gases and leach harmful substances into the environment, PCDR emerges as a highly competitive and sustainable solution. Additionally, PCDR presents an opportunity to utilize carbon dioxide, aiding in climate change mitigation and promoting circular economy principles.

However, despite its potential, the technology is still in its early stages and requires further research before industrial-scale implementation. Key research area include suitable catalysts to ensure appropriate efficiency and stability of the process. Ni- or Co- based oxide catalysts commonly used in reforming processes are not sufficiently active or stable (Chen et al., 2020; Pawar et al., 2015; Pawelczyk et al., 2022). Transition metal carbides, in particular Mo₂C, have emerged as promising alternatives, exhibiting favorable activity in hydrocarbon reforming, and enhanced stability due to their characteristic oxidation-re carburization cycles (Czaplicka et al., 2021; Ma et al., 2017; Wang et al., 2022). Moreover, Mo₂C exhibits high catalytic activity in reforming of hydrocarbons with longer carbon chains, which is a promising prospect considering that the diverse hydrocarbon mixture is reformed in the PCDR process, not limited to methane (Czaplicka et al., 2021; Pawelczyk et al., 2022). Therefore, combining Ni or Co with carbide phase may result in beneficial effects on the catalyst performance.

The published research in the field of catalysts for PCDR is narrow, mostly limited to the conventional Ni or Co supported on oxides catalysts (Pawelczyk et al., 2022). Therefore, this work reveals novel Ni-Mo₂C/ γ -Al₂O₃ catalysts. In order to obtain Mo₂C, the temperature-programmed carburization of organic-inorganic precursor method was used, offering a safer and more energy-efficient alternative to conventional methods that require utilization of gas hydrocarbons and hydrogen. Various amines (aliphatic, cyclic, aromatic) were used to synthesize the catalyst, demonstrating their influence on structural properties and activity. Catalyst characterization included BET surface area, XRD, CO₂-TPD, H₂-TPR, SEM and TGA. Catalytic activity was examined in PCDR of model mixture of polymers (LDPE, HDPE, PP, PS and PET) representing municipal waste plastics. The setup comprised a pyrolysis reactor followed by a fixed bed reforming reactor. Catalyst-to-plastic mass ratio was 0.5. The catalytic testing comprises the catalytic dry reforming of gases generated during pyrolysis-gasification step. The reforming was performed at 800°C, while the pyrolysis step was conducted from room temperature to 500°C. Output gases were chilled to capture condensable products, while remaining gases were analysed using gas chromatography. Non-catalytic PCDR served as a reference. Catalyst performance was evaluated based on synthesis gas yield, H₂:CO ratio, product distribution, CO₂ conversion, and catalyst stability. XRD, SEM, and TG analyses were also conducted post-reaction to assess structural changes and catalysts deactivation.

As presented in Fig. 1a, Ni-Mo₂C/ γ -Al₂O₃_A, Ni-Mo₂C/ γ -Al₂O₃_C, and Ni-Mo₂C/ γ -Al₂O₃_H catalysts, synthesized using different amines (aniline, cyclohexylamine and hexylamine, respectively) showed high activity, resulting in notable improvements in syngas yields compared to non-catalytic process. The addition of catalysts enabled a syngas yield increase of over three times and a hydrogen yield boost of roughly tenfold. Moreover, different amines used for synthesis, influenced obtained syngas yield and composition, with Ni-Mo₂C/ γ -Al₂O₃_H exhibiting the highest yields and H₂/CO ratio. In addition, obtained product distributions varied for different catalysts, indicating that variety of value-added products can be recovered from waste, depending on the catalyst

used. The investigation has shown that amine, besides carbon source, is a structure-forming factor that has a significant impact on the properties of the catalyst and, consequently, on its activity and stability. Catalyst properties, including surface area, pore size, basicity, and structure were affected. Catalysts also varied in crystallite size and Mo phase presence (Fig. 1b).

The obtained results indicate that catalysts based on metal carbides are promising candidates as catalysts for syngas generation from waste plastics. A significant influence of amine choice for catalyst synthesis was demonstrated. It was revealed that Ni-Mo₂C/γ-Al₂O₃ synthesized using hexylamine is the most promising catalyst among tested catalysts. Moreover, the carbide synthesis method is competitive to conventional methods in terms of cost and safety. Therefore, this work constitutes an important contribution to the field regarding pyrolysis combined with dry reforming as a sustainable method of plastic waste management.

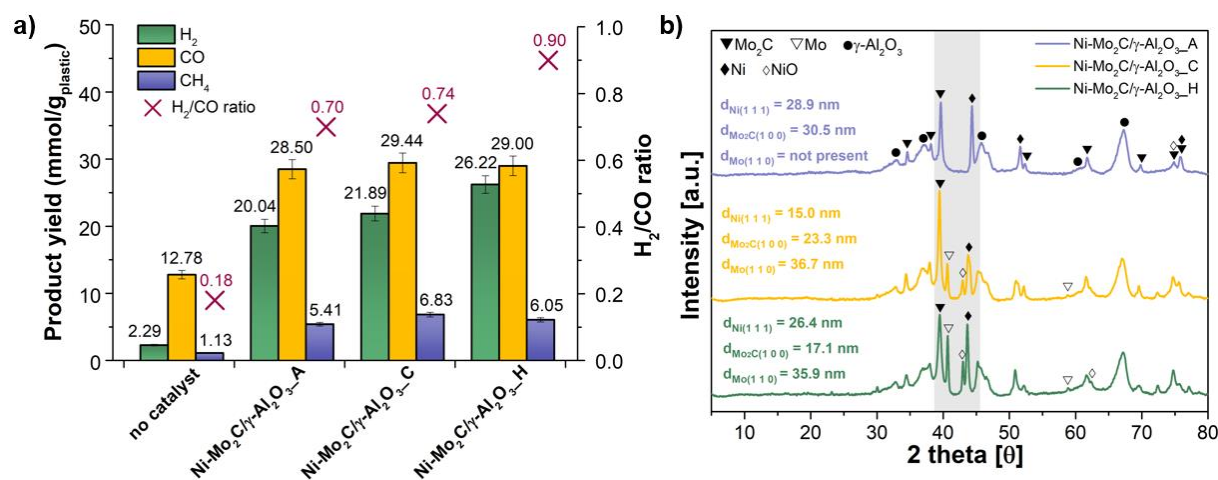


Fig. 1 a) Product yields and H₂/CO molar ratio obtained in PCDR of model mixture of plastics carried out without catalyst and over Ni-Mo₂C/γ-Al₂O₃ catalysts, b) XRD patterns and crystallite sizes for the synthesized Ni-Mo₂C/γ-Al₂O₃ catalysts.

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